- Which compound is **not** a fertiliser?
 A ammonium sulfate, (NH₄)₂SO₄
 B calcium hydroxide, Ca(OH)₂
 C potassium chloride, KC*l*
 - **D** urea, $CO(NH_2)_2$
- 2 Acid rain is formed when sulfur dioxide and oxides of nitrogen dissolve in rain water.

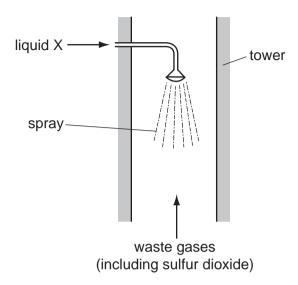
Which problem is not caused by acid rain?

- A breathing difficulties
- **B** dying trees
- C erosion of statues
- **D** lowered pH of lakes
- 3 Statement 1: The burning of fossil fuels containing sulfur is a cause of 'acid rain'.
 - Statement 2: Acid rain contains sulfur dioxide which is formed when sulfur compounds burn in the air.

Which of the following is true?

- **A** Both statements are correct and statement 2 explains statement 1.
- **B** Both statements are correct, but statement 2 does not explain statement 1.
- **C** Statement 1 is correct but statement 2 is incorrect.
- **D** Statement 2 is correct but statement 1 is incorrect.

4 When coal and oil burn in power stations, the acidic gas sulfur dioxide is formed. Sulfur dioxide is removed by absorbing it in a liquid sprayed down a tower.



What is liquid X?

- A calcium hydroxide solution
- B sodium chloride solution
- C dilute hydrochloric acid
- **D** water